

PRN NO

QP. CODE

Dr. J. J. Magdum Trust's  
**Dr. J. J. Magdum College of Engineering, Jaysingpur**  
 (An Autonomous Institute)  
**Test I Sem – I A.Y. – 2024-25**

Class Program	FYBTech (all divisions)
Course Code	01FYBSL101
Course Title	Applied Mathematics – I

Day & Date	Saturday, 26/10/2024
Time	10.00 am to 11.00 am
Max. Marks	20

Q.No		Solve the following.	Marks	CO	BL	PI
1	a	Reduce the following matrix to normal form and find its rank. $\begin{bmatrix} 1 & 1 & 1 & -1 \\ 1 & 2 & 3 & 4 \\ 3 & 4 & 5 & 2 \end{bmatrix}$	05	1,3	L1, L3	1.1.1
	b	Test for consistency and if possible, solve the equations $2x - y + z = 9, \quad 3x - y + z = 6, \quad 4x - y + 2z = 7,$ $-x + y - z = 4$	05	1,3	L2, L3	1.1.1
	<b>OR</b>					
b	Investigate for what value of $\lambda$ and $\mu$ the equations $x + y + z = 6, \quad x + 2y + 3z = 10, \quad x + 2y + \lambda z = \mu.$ have i) no solution ii) unique solution					
2	a	Find Eigen values of the matrix A and also find eigen values of (i) $A^{-1}$ (ii) adjoint A where $A = \begin{bmatrix} 1 & 2 & 3 \\ 0 & -2 & 6 \\ 0 & 0 & -3 \end{bmatrix}$	05	1,2	L2, L3	2.4.1
	<b>OR</b>					
	a	Find Eigen values and eigen vector for largest eigen value of the matrix $A = \begin{bmatrix} 8 & -8 & -2 \\ 4 & -3 & -2 \\ 3 & -4 & 1 \end{bmatrix}$				
	b	Solve the following homogeneous equations $4x + 3y - z = 0, \quad 3x + 4y + z = 0, \quad 5x + y - 4z = 0$	05	1,3	L2, L3	1.1.1

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TA105

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**Dr. J. J. Magdum College of Engineering, Jaysingpur**  
 (An Autonomous Institute)  
**Test II Sem – I A.Y. – 2024-25**

Class Program	F.Y.B.Tech.
Course Code	01FYBSL101
Course Title	Applied Mathematics – I

Day & Date	Saturday, 07/12/2024
Time	01.00 pm to 02.00 pm
Max. Marks	20

Q.No		Solve the following.	Marks	CO	BL	PI
1	a	Simplify: $\frac{(\cos 3\theta + i \sin 3\theta)^5 (\cos \theta - i \sin \theta)^3}{(\cos 5\theta + i \sin 5\theta)^7 (\cos 2\theta - i \sin 2\theta)^5}$	05	2	L1, L2 L4	1.1.1 3.1.1
	b	Find all the values of $(1 - i)^{2/3}$	05	2	L1 L2, L4	1.1.1 3.1.1
	<b>OR</b>					
	b	Solve the equation $x^4 + x^3 + x^2 + x + 1 = 0$				
2	a	Arrange in powers of x by using Taylor's series $17 + 6(x + 2) + 3(x + 2)^3 + (x + 2)^4 - (x + 2)^5$	05	2	L1 L2, L3	2.3.1
	<b>OR</b>					
	a	Using Taylor's series expand $\sin\left(\frac{\pi}{6} + x\right)$ upto $x^4$ and find $\sin(30^\circ, 30')$ .				
	b	Using De Moivre's Theorem Prove that $\frac{\sin 6\theta}{\sin 2\theta} = 16\cos^4\theta - 16\cos^2\theta + 3$	05	2	L1 L2, L4	1.1.1 3.1.1

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TA103

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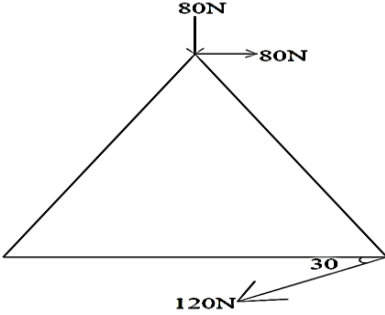
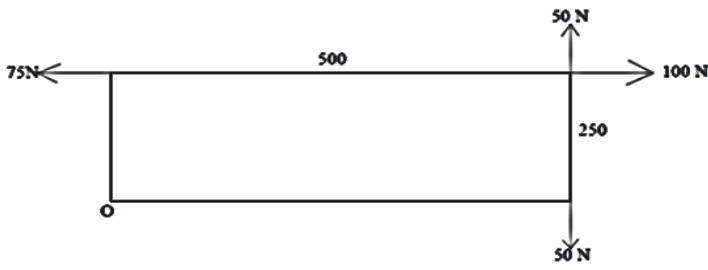
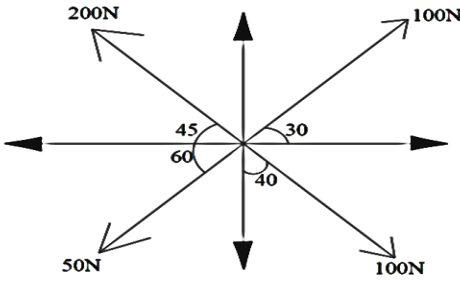
**Dr. J. J. Magdum College of Engineering, Jaysingpur**

(An Autonomous Institute)

**Test I / II Sem – I A.Y. – 2024-25**

Class Program	AI&DS/ CSE/ IT
Course Code	01FYESL107
Course Title	Applied Mechanics

Day & Date	Friday, 25/10/2024
Time	10am- 11am
Max. Marks	20

Q. No	Question	Marks	CO	BL	PI	
1	a State "Lami's Theoram" and Explain with neat sketch.	05	CO2	Remember	1.3.1	
	b State "Law of Parallelogram of Forces" and Draw a neat sketch.	05	CO1	Remember	1.3.1	
	OR			Remember		
b	State the Law of transmissibility of forces and draw the neat sketch of it.					
2	a Determine the resultant of non-concurrent force system acting on the lamina of equilateral triangular shape as shown in figure	05	CO1	Understand	2.3.1	
						
	OR					
a	Determine the resultant of non-concurrent force system as shown in figure			Understand		
						
b	A system of four forces acting on a body as shown in figure. Determine the resultant.	05	CO1	Understand	2.3.1	
						

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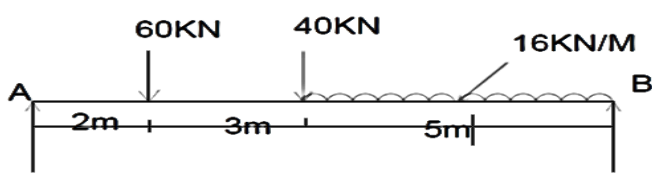
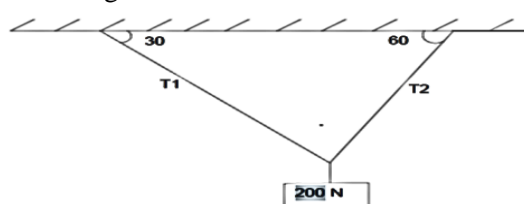
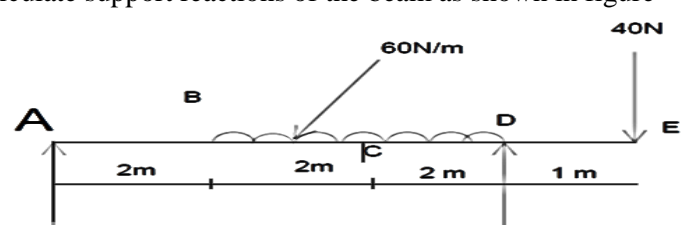
**Dr. J. J. Magdum College of Engineering, Jaysingpur**

(An Autonomous Institute)

**Test II Sem – I A.Y. – 2024-25**

Class Program	FYBTECH
Course Code	01FYESL107
Course Title	Applied Mechanics

Day & Date	Friday & 06/12/2024
Time	02:00 pm- 03:00 pm
Max. Marks	20

Q. No	Attempt the following	Marks	CO	BL	PI
1	a Define free body diagram with neat sketch. Give two example.	05	CO2	Remember	1.1.a
	b State & explain the different types of support with neat sketch.	05	CO2	Understand	1.3.1
	OR				
	b State & explain Parallel axis theorem with neat sketch.		CO3	Understand	1.2.1
2	a Calculate support reactions of the beam as shown in figure 	05	CO2	Apply	2.1.3
	OR				
	a A weight of 200N is attached by two strings. Calculate the tension in the strings. 		CO2	Apply	2.1.3
b Calculate support reactions of the beam as shown in figure 	05	CO2	Apply	2.1.3	

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TA102

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**Dr. J. J. Magdum College of Engineering, Jaysingpur**  
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**Test I Sem – I A.Y. – 2024-25**

Class Program	FY CSE
Course Code	01FYESL105
Course Title	Basic Electronics Engineering

Day & Date	Thursday
Time	2.00pm to 3.00pm
Max. Marks	20

Q.No			Marks	CO	BL	PI
1	a	How transistor can be used to form an amplifier? Explain with circuit diagram and waveform.	05	CO1	Understand	
	b	What are different number systems? Define radix of number system. Mention it for each of the number system.	05	CO2	Understand, Apply	
	OR					
b	In the following binary number $(1001011011010)_2$ Write: i) Decimal weight of LSB ii) Decimal weight of MSB iii) Number of nibbles iv) Number of bytes v) Number of bits					
2	a	Plot & explain the DC load line on output characteristics of CE configuration of transistor. Define Q/Operating point	05	CO1	Understand	
	OR					
	a	Explain the working of bridge rectifier with circuit diagram and waveform.				
	b	Convert following codes- i) $(2AF7)_{16} = (?)_2$ ii) $(764)_{10} = (?)_{16}$	05	CO2	Understand, Apply	

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**Test I / II Sem – I A.Y. – 2024-25**

Class Program	FY BTech
Course Code	01FYESL105
Course Title	Basic Electronics Engineering

Day & Date	06/12/2024
Time	10:00AM To 11:00
Max. Marks	20

Q.No		Marks	CO	BL	PI	
1	a	Convert the following: i) $(673)_8 = (?)_{10}$ ii) $(9543)_{10} = (?)_{BCD}$	05	2	L3	
	b	Explain Half subtractor with Definition, truth table, K-map, logic equation and logic diagram.	05	4	L2	
	OR					
	b	Explain Half adder with Definition, truth table, K-map, logic equation and logic diagram.				
2	a	State and explain De-Morgan's theorems.	05	3	L2	
	OR					
	a	Explain AND gate, OR gate and NOT with statement, logic diagram, equation and truth table.				
	b	Justify the statement "NAND and NOR gates are known as universal gates". Draw Ex-OR and Ex-Nor gate using NAND gate.	05	3,4	L2	

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**Test I / II Sem – I A.Y. – 2024-25**

Class Program	F.Y.Civil/Mechanical/ETC
Course Code	01FYESL104
Course Title	Basic Electrical Engineering

Day & Date	Thursday-24/10/2024
Time	3.30 pm to 4.30 pm
Max. Marks	20

Q.No		Marks	CO	BL	PI	
1	a	Explain effect of temperature on resistance material.	05	1	L2	K1
	b	Explain and State Kirchhoff's Current Law..	05	1	L2	K1
	OR					
	b	Explain and State Kirchhoff's Voltage Law.				
2	a	Two Batteries A & B are Connected in Parallel across a Load Resistance of $4 \Omega$ . The EMF and internal resistance of battery A and B are 24 V, $4 \Omega$ and 36 V, $6 \Omega$ respectively, using mesh analysis Find current in Battery A and current in Battery B.	05	2	L3	K3
	OR					
		a	Two Batteries A & B are Connected in Parallel across a Load Resistance of $6 \Omega$ . The EMF and internal resistance of battery A and B are 32 V, $4 \Omega$ and 36 V, $6 \Omega$ respectively, using node analysis Find current in Battery A and current in Battery B.			
	b	Define (i) Magnet, (ii) Magnetic Field, (iii) Magnetic Flux, (iv) Flux density.	05	1	L2	K1

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**Test - II Sem – I A.Y. – 2024-25 Div - D, E & F**

Class Program	FYBTech
Course Code	FYESL104
Course Title	Basic Electrical Engineering

Day & Date	Friday 06/12/2024
Time	11:30 am to 12:30 pm
Max. Marks	20

Q. No		Marks	CO	BL	PI	
1	a	Comparing between electric circuit and magnetic circuit.	05	1	K1	1.2.1
	b	Explain Magnetic leakage and fringing.	05	1	K1	1.2.1
	OR			1	K1	1.2.1
	b	Derive the expression of series magnetic circuit for three different magnetic material.				
2	a	Derive the expression of RMS value of alternating current of sine wave.	05	3	K4	2.1.1
	OR			3	K4	2.1.1
	a	Derive the equation of current and power in pure inductive circuit				
	b	A R=50 ohm, L=100 mH, C=50 microfarad are connected in series across 230 volts, 50Hz ac supply find i) Impedance      ii) Current      iii) Power factor .	05	2	K3	2.1.2



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TA108

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**Test I / II Sem – I A.Y. – 2024-25**

Class Program	Mech/ Civil/ E&TC Engineering
Course Code	01FYESL106
Course Title	Computer Aided Engineering Drawing

Day & Date	Friday, 25/10/2024
Time	11.30-12.30
Max. Marks	20

Q. No		Question	Marks	CO	BL	PI
1	a	Enlist types of lines and illustrate with neat diagram (Any 5)	05	CO1	Remember	1.3.1
	b	Write a note on page set for AutoCAD 2D drawing.	05	CO2	Remember	5.1.1
	OR					
	b	Write the path for following draw commands 1. Circle 2. Polygon 3. Construction line 4. Hatch 5. Text				
2	a	Draw BIS sign conventions for any 5 materials.	05	CO1	Analyze	1.3.1
	OR					
	a	What are the types of dimensioning system? explain in brief with neat sketch.			Understand	1.3.1
	b	Write the path for following modify commands 1. Trim 2. Mirror 3. Move 4. Offset 5. Array	05	CO2	Remember	5.1.1

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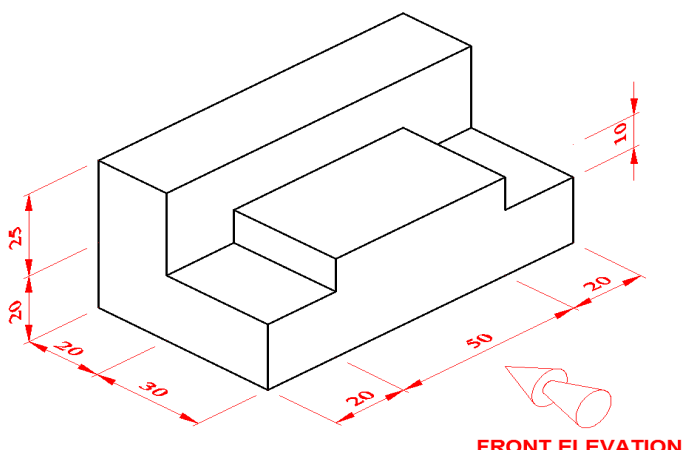
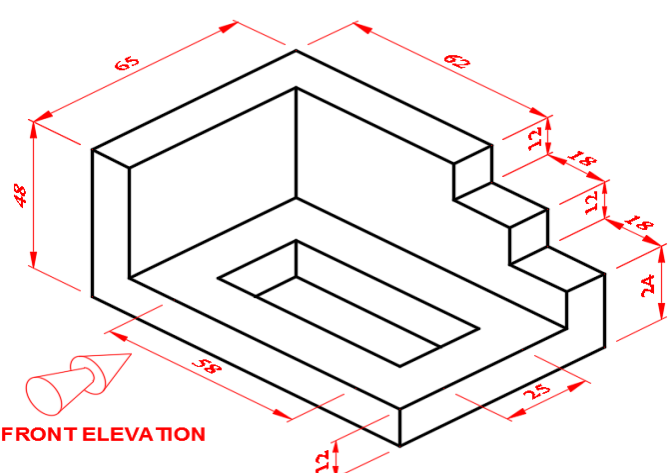
**Dr. J. J. Magdum College of Engineering, Jaysingpur**

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**Test I / II Sem – I A.Y. – 2024-25**

Class Program	FYBTECH
Course Code	01FYESL106
Course Title	Computer Aided Engineering Drawing

Day & Date	Friday, 31/12/2024
Time	1:00- 2:00 pm
Max. Marks	20

Q.No		Question	Marks	CO	BL	PI
1	a	Differentiate between first and third angle method	04	CO4	Remember	1.3.1
	b	 <p>FRONT ELEVATION</p> <p>Draw orthographic projections by <b>Third Angle</b> method</p>	06	CO4	Apply	2.1.3
2	a	 <p>FRONT ELEVATION</p> <p>Draw Orthographic Projections By <b>First angle</b> method</p>	10	CO4	Apply	2.1.3
	b	<p>or</p> <p>A pentagonal lamina of 40 mm side has a circular hole of 15 mm diameter in its centre. The plane stands on one of its sides on HP with its plane perpendicular to VP and <math>45^\circ</math> inclined to HP. Draw the projections.</p>	10	CO3	Apply	2.1.3

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(An Autonomous Institute)

**Test II Sem – I A.Y. – 2024-25 Div- C**

Class Program	FYBTech	Day & Date	Thursday, 7 <sup>th</sup> Dec
Course Code	01FYBSL103	Time	10.00 to 11.00 am
Course Title	Engineering Chemistry	Max. Marks	20

Q.No			Marks	CO	BL	PI
1	a	Define electrochemical corrosion. Explain oxygen absorption mechanism with example.	05	03	02	2.1.3
	b	What is hot dipping process? Discuss galvanisation with a neat labelled diagram.	05	01	01	1.2.1
	OR					
	b	What is hot dipping process? Discuss tinning with a neat labelled diagram.				
2	a	Write a note on Cathodic protection.	05	02	02	1.2.1
	OR					
		a	Write a note on Electroplating..			
	b	A Bomb calorimeter experiment gave the following data : i) Water equivalent of calorimeter = 550 gm ii) Weight of coal sample = 1.1 gm iii) Weight of water taken = 2460 gm iv) Observed rise in temperature = 2.51OC v) Cooling correction = 0.043 OC vi) Correction due to H <sub>2</sub> SO <sub>4</sub> = 24.2 Cal vii) Correction due to HNO <sub>3</sub> = 35.8 Cal viii) Fuse wire correction = 4.0 Cal If the fuel contains 5% hydrogen, calculate the Gross and Net calorific value of the fuel. (Given latent heat of steam = 587 Cal/gm).	05	03	02	2.1.3

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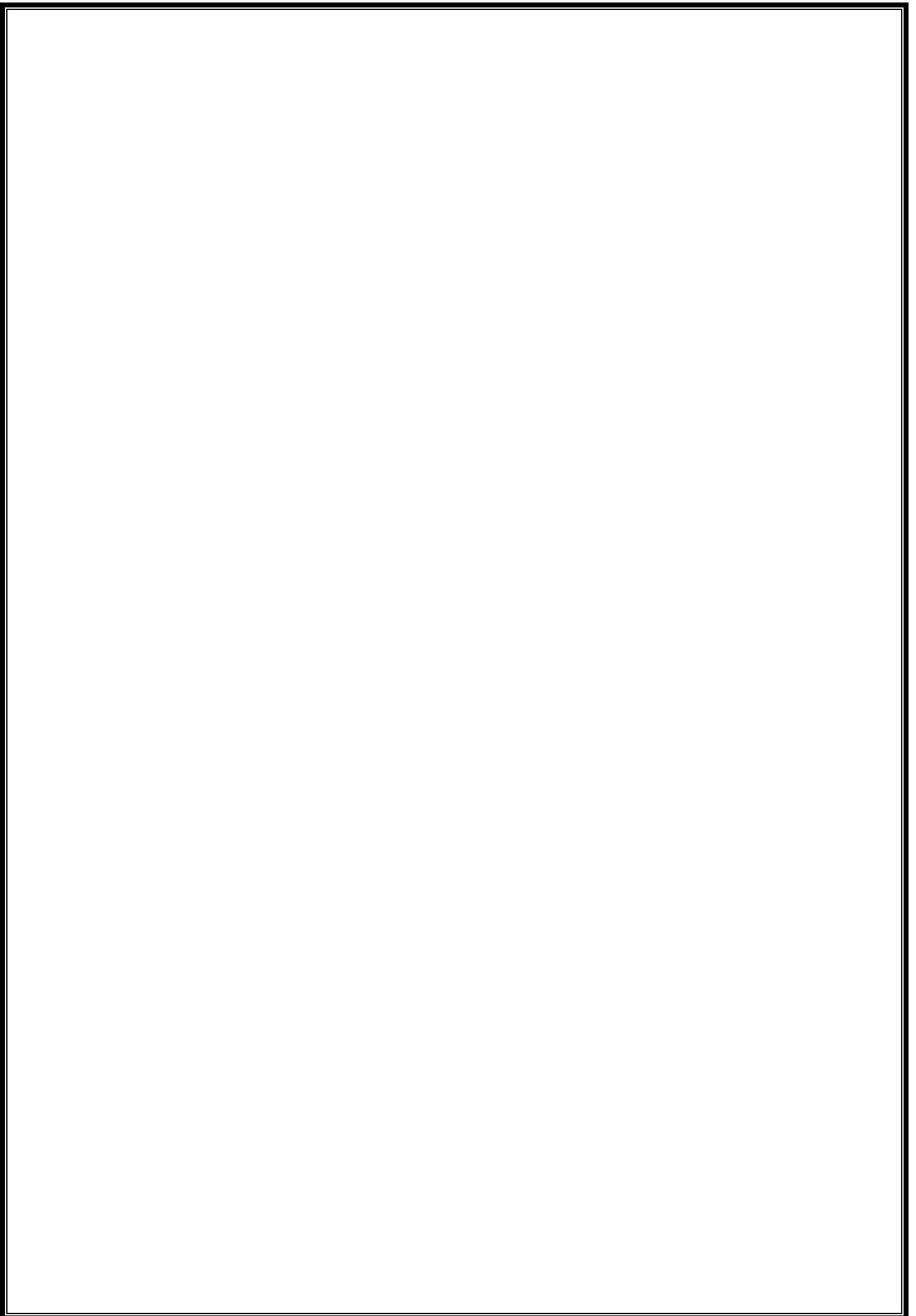
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 (An Autonomous Institute)  
**Test II Sem – I A.Y. – 2024-25 Div- A & B**

Class Program	FYBTECH
Course Code	01FYBSL103
Course Title	Engineering Chemistry

Day & Date	07 Dec, 2024
Time	10.00 – 11.00 am
Max. Marks	20

Q. No		Marks	CO	BL	PI															
1	a	05	03	02	2.1.3															
	A Bomb calorimeter experiment gave the following data :																			
	i) Water equivalent of calorimeter = 550 gm ii) Weight of coal sample = 1.1 gm iii) Weight of water taken = 2460 gm iv) Observed rise in temperature = 2.51OC v) Cooling correction = 0.043 OC vi) Correction due to H <sub>2</sub> SO <sub>4</sub> = 24.2 Cal vii) Correction due to HNO <sub>3</sub> = 35.8 Cal viii) Fuse wire correction = 4.0 Cal If the fuel contains 5% hydrogen, calculate the Gross and Net calorific value of the fuel. (Given latent heat of steam = 587 Cal/gm)																			
	b					Write a brief note on- Alkalinity of water.														
	OR		05	01	01	1.2.1														
	b	Write a brief note on- Chloride content of water .																		
2	a	05	02	02	1.2.1															
	Discuss in detail about the scale & sludge formation in boilers.																			
	OR																			
	a					Explain the Ion Exchange process for water softening.														
	b	05	03	02	2.1.3															
A sample of water on analysis was found to contain the following impurities																				
<table style="width: 100%; border-collapse: collapse;"> <tr> <td style="width: 30%;"></td> <td style="width: 20%; text-align: center;">Wt. mg/lit</td> <td style="width: 20%; text-align: center;">Mol. wt.</td> </tr> <tr> <td>CaCO<sub>3</sub></td> <td style="text-align: center;">12.5</td> <td style="text-align: center;">100</td> </tr> <tr> <td>Mg(HCO<sub>3</sub>)<sub>2</sub></td> <td style="text-align: center;">20.6</td> <td style="text-align: center;">146</td> </tr> <tr> <td>CaCl<sub>2</sub></td> <td style="text-align: center;">11.1</td> <td style="text-align: center;">111</td> </tr> <tr> <td>MgSO<sub>4</sub></td> <td style="text-align: center;">7.8</td> <td style="text-align: center;">120</td> </tr> </table>							Wt. mg/lit	Mol. wt.	CaCO <sub>3</sub>	12.5	100	Mg(HCO <sub>3</sub> ) <sub>2</sub>	20.6	146	CaCl <sub>2</sub>	11.1	111	MgSO <sub>4</sub>	7.8	120
	Wt. mg/lit					Mol. wt.														
CaCO <sub>3</sub>	12.5	100																		
Mg(HCO <sub>3</sub> ) <sub>2</sub>	20.6	146																		
CaCl <sub>2</sub>	11.1	111																		
MgSO <sub>4</sub>	7.8	120																		
Calculate temporary, permanent and total hardness of water in ppm.																				



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**Test I / II Sem – I A.Y. – 2024-25**

Class Program	FYBTech
Course Code	01FYBSL103
Course Title	Engineering Chemistry

Day & Date	Thursday, 24/10/2024
Time	10:00- 11.00 am
Max. Marks	20

Q.No		Marks	CO	BL	PI
1	a	Derive the expression for Beer-Lambert's Law.	05	03	02
	b	Compare Solid fuels with Liquid fuels.	05	01	01
	OR				
	b	Compare Liquid fuels with Gaseous fuels.			
2	a	Discuss the principal and working of Bomb Calorimeter.	05	02	02
	OR				
	a	Discuss the principal and working of Boy's Calorimeter.			
	b	With the schematic diagram, explain the principle & working of Single Beam Spectrophotometer. How will you determine the concentration of unknown solution with the help of Single Beam Spectrophotometer?	05	03	02

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**Test II Sem-I A.Y. - 2024-25**

<b>Class Program</b>	FYBTECH
<b>Course Code</b>	01FYESL108
<b>Course Title</b>	Fundamentals of Programming Language

<b>Day &amp; Date</b>	Friday: 07/12/2014
<b>Time</b>	3.00 pm to 4.00 Pm
<b>Max. Marks</b>	20

Que. No	#	Questions	Marks	CO	BL	PI
1.	a.	Define Arrays in C with declaration & initialization.	05	CO3	L1	1.3.1
	b.	Explain if else-if ladder and Nested if else with example.	05	CO2	L2	1.4.1
	<b>OR</b>					
	b.	Write a C program to display even numbers and odd numbers in between 1-100 using For Loop.	05	CO2	L6	1.4.1
2.	a.	Explain For Loop and While Loop with example.	05	CO2	L2	1.4.1
	<b>OR</b>					
	a.	Explain difference between Arrays and String.	05	CO3	L2	1.4.1
	b.	Write a C program to display week-days using Switch statement.	05	CO2	L6	1.4.1

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**Test I****Sem – I A.Y. – 2024-25**

Class/ Program	Div.D/E/F
Course Code	01FYBSL102
Course Title	Engineering Physics

Day & Date	Thursdy,24/10/2024
Time	11:30-12:30 pm
Max. Marks	20

Q.No		Marks	CO	BL	PI
1	a	Define resolving power of grating and obtain an expression for it.	06	1,2	K <sup>1</sup> , K <sup>2</sup>
	OR				
	a	What is grating? Explain theory of plane transmission grating when light falls at 90° on the grating.			
	b	A parallel beam of Sodium light is allowed to be incident normally on a plane grating having 4250 lines per cm and a second order spectral line is observed to be deviated through 30°. Calculate the wavelength of spectral line.	04	3	K <sup>3</sup>
2	a	Define: optic axis, Principal plane. With neat diagram explain the phenomenon of double refraction.	06	1,2	K <sup>1</sup> , K <sup>2</sup>
	OR				
	a	Give Huygen's theory of double refraction in uniaxial crystal & distinguish between positive and negative crystals.	04	1	K <sup>1</sup>
	b	Define any four terms from the following: Diffraction, Polarised light, anisotropic media, principal section, principal plane.			



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**Test II****Sem – I A.Y. – 2024-25**

Class/ Program	FYBTECH
Course Code	01FYBSL102
Course Title	Engineering Physics

Day & Date	Saturday, 07/12/2024
Time	11.30 am- 12.30 pm
Max. Marks	20

Q.No		Marks	CO	BL	PI	
1	a	Explain the following terms( <b>Any two</b> ): i)Spontaneous Emission ii)Stimulated Emission iii)Population inversion	06	1,2	K <sup>1</sup> , K <sup>2</sup>	1.2.1
	OR					
	a	What is holography? Explain recording of hologram and reconstruction of hologram.				
	b	What do you mean by pumping& Explain optical pumping technique & electric discharge pumping technique.	05	1,2	K <sup>1</sup> , K <sup>2</sup>	1.2.1
2	a	What is acoustics of buildings? State the essential features of good acoustics?	06	1,2	K <sup>1</sup> , K <sup>2</sup>	1.2.1
	OR					
	a	Explain the term reverberation. State and explain factors controlling reverberation time. State and explain sabine's formula.				
	b	An auditorium has a volume of 8000 m <sup>3</sup> .It is required to have reverberation time of 2 seconds. What should be the total absorption in it?	03	3	K <sup>3</sup>	2.1.3